



Europäisches Patentamt
European Patent Office
Office européen des brevets



(11) **EP 1 418 255 A1**

(12) **EUROPEAN PATENT APPLICATION**

(43) Date of publication:
12.05.2004 Bulletin 2004/20

(51) Int Cl.7: **C25F 5/00**

(21) Application number: **03256301.7**

(22) Date of filing: **07.10.2003**

(84) Designated Contracting States:
**AT BE BG CH CY CZ DE DK EE ES FI FR GB GR
HU IE IT LI LU MC NL PT RO SE SI SK TR**
Designated Extension States:
AL LT LV MK

(72) Inventors:
• **Kryzman Michael A.
West Hartford ,CT 06107 (US)**
• **Jarowski,Mark R.
Glastonbury,CT 06033 (US)**

(30) Priority: **09.10.2002 US 268337**

(74) Representative: **Leckey, David H.
Frank B. Dehn & Co.,
179 Queen Victoria Street
London EV4V 4EL (GB)**

(71) Applicant: **United Technologies Corporation
Hartford Connecticut 06101 (US)**

(54) **Electrochemical Process for the Simultaneous Stripping of Diverse Coatings From a Metal Substrate**

(57) An electrochemical process for simultaneously stripping diverse coatings from a metal substrate and, more particularly to the removal of MCrAlY and alumide coatings from a base metal comprises immersing

the substrate in an electrolyte and applying a potential across the electrolyte which is sufficiently large to dissolve and remove the coatings.

EP 1 418 255 A1

Description

BACKGROUND OF THE INVENTION

[0001] The present invention relates to an electrochemical process for simultaneously stripping diverse coatings from a metal substrate and, more particularly to the removal of MCrAlY and aluminide coatings from a base metal.

[0002] Elements of gas turbine engines are protected from high oxidation and corrosion by coating the base metals with a protective coating. Gas turbine engine combustor float wall elements are protected from high temperature oxidation and corrosion with two types of coatings. The first coating is a coating of MCrAlY on the inside surface of a combustor float wall and an aluminide coating on the outside surface of the float wall. To date, the repair sequence for such elements requires the removal of coatings from the base metal. The current process involves separate stripping techniques for the two coatings. The MCrAlY coating is removed either by soaking the parts in a high concentrated hot hydrochloric acid solution or by water jet blasting. The aluminide coating is removed by stripping in nitric acid. These processes are difficult to control, are hazardous, and extremely labor intensive.

[0003] It is clear that indeed remains for an approved process for stripping diverse coatings from a metal substrate.

[0004] It is therefore the primary object of the present invention to provide such a process.

SUMMARY OF THE INVENTION

[0005] The process of the present invention allows for the simultaneous removal of at least two diverse coatings from the metal substrate. The metal substrate having the at least two diverse coatings is immersed in an electrolyte and then a potential is applied across the electrolyte at a magnitude sufficient to dissolve the at least two coatings and remove them from the metal substrate. The process is particularly useful for removing diverse coatings of M chrome aluminum yttrium MCrAlY (where capital letter M is nickel and/or cobalt) and aluminide coatings from a metal substrate.

[0006] Further features of the present invention will appear hereinbelow.

BRIEF DESCRIPTION OF THE DRAWINGS

[0007] A detailed description of preferred embodiments of the present invention follows, with reference to the attached drawings, wherein:

Figures 1a, b and c schematically illustrate a process in accordance with the present invention and Figure 2 illustrates an electrochemical record of the electrochemical process for the simultaneous strip-

ping of diverse coatings in accordance with the present invention.

DETAILED DESCRIPTION

[0008] The present invention relates to an electrochemical process for simultaneously stripping diverse coatings from the metal substrate and, more particularly, to the removal of MCrAlY (where M is nickel and/or cobalt) and aluminide coatings from a base metal.

[0009] In accordance with the present invention, the base metal is typically any alloy suitable for use in high temperature oxidation and corrosion environments. Suitable base metal alloys include stainless steel alloys, nickel base alloys cobalt base alloys and the like. The base metal substrate is provided with diverse coatings. In the particular high temperature oxidation and corrosion environments for which the base metals are employed, diverse coatings are often required. For example, in the case of combustor float walls used in gas turbine engines, the base metals are coated with a high temperature resistant metal coating and an aluminide coating. Typically, the inside surface of the metal substrate is coated with MCrAlY (where M is nickel and/or cobalt) the outside surface of the engine combustor float wall is coated with an aluminide coating.

[0010] It has been found in accordance with the process of the present invention that the diverse coatings on the metal substrate can be simultaneously removed by employing an electrochemical process for stripping the diverse coatings from the metal substrate. The metal substrate having at least two diverse coatings thereon is immersed in an electrolyte and a potential is applied across the electrolyte at sufficient magnitude to dissolve and remove the two diverse coatings from the metal substrate.

[0011] It has been found that a suitable electrolyte comprises an acid solution. Suitable acid solutions include hydrochloric acid, nitric acid and sulfuric acid. However, hydrochloric acid is preferred as it acts faster than either nitric acid or sulfuric acid. The preferred electrolyte comprises a 5 to 10 volume percent solution of hydrochloric acid.

[0012] In accordance with a preferred embodiment of the present invention, the electrolyte is agitated while applying the potential. In addition, the process is preferably carried out under ambient conditions. The potential applied in accordance with the present invention is greater than +50 mV and up to about +150 mV volts versus an Ag/AgCl reference electrode.

[0013] As noted above, the base metal substrate includes any high temperature corrosion resistant alloy including stainless steels, nickel base alloys, nickel and cobalt based alloys, and the like.

[0014] Figure 1 schematically illustrates a process in accordance with the present invention. As shown in Figure 1, a suitable vessel 10 is provided, and the diversely coated metal substrate element 12 to be treated as po-

sitioned therein. The element 12 is preferably positioned between cathodes 14, 16 which may advantageously be graphite cathodes or made of other materials with high corrosion resistance to mineral acids, e.g. Hastelloy C-22, and a reference electrode 18 is positioned extending into an electrolyte solution.

[0015] The element 12 may advantageously be suspended in the solution contained within the vessel 10, and structures used to suspend the element 12 should be selected from a material which will not be affected by the conditions in material within the vessel 10. For example, in accordance with the present invention, titanium wire is particularly suitable for securing element 12 as desired.

[0016] As set forth above, it may be desirable to agitate the electrolyte within the vessel 10 and this may be accomplished, for example, by using any suitable mixing or agitation devices would be readily known to a person skilled in the art.

[0017] It is to be understood that the invention is not limited to the illustrations described and shown herein, which are deemed to be merely illustrative of a preferred embodiment of the invention, and which are susceptible of modification of form, size, arrangement of parts and details of operation. The invention rather is intended to encompass all such modifications which are within its scope as defined by the claims.

Claims

1. A process for the simultaneous removal of at least two diverse coatings from a metal substrate comprising the steps of:
 - providing an element (12) comprising a metal substrate having at least two diverse coatings thereon;
 - immersing the element in an electrolyte; and
 - applying a potential across the electrolyte at a magnitude wherein the at least two coatings are dissolved and removed from the metal substrate.
2. A process according to claim 1 wherein said electrolyte comprises an acid solution.
3. A process according to claim 2 wherein the acid solution is selected from the group consisting of hydrochloric acid solution, nitric acid solution, sulfuric acid solution and mixtures thereof.
4. A process according to claim 2 wherein the acid solution is a hydrochloric acid solution.
5. A process according to claim 4 the electrolyte comprises 5 to 10 volume % solution of hydrochloric acid.
6. A process according to any preceding claim including agitating the electrolyte while applying the potential.
7. A process according to any preceding claim including carrying out the process under ambient conditions.
8. A process according to any preceding claim wherein the metal substrate is coated with MCrAlY on one surface (where M is selected from the group consisting of Ni, Co, and mixtures thereof) and aluminate on the other surface thereof.
9. A process according to any preceding claim wherein the metal substrate is a metal alloy selected from the group consisting of stainless steel alloys, nickel alloys, cobalt alloys, and nickel-cobalt alloys.
10. A process according to any preceding claim wherein the potential is greater than +50 mV and up to about +150 mV volts versus a Ag/AgCl reference electrode.

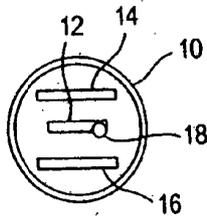


FIG. 1a

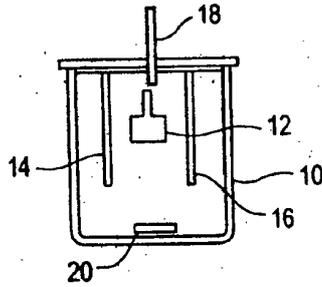


FIG. 1b

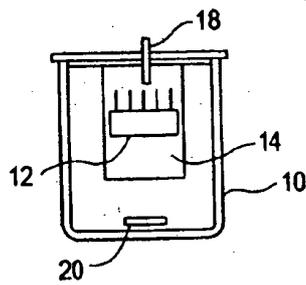


FIG. 1c

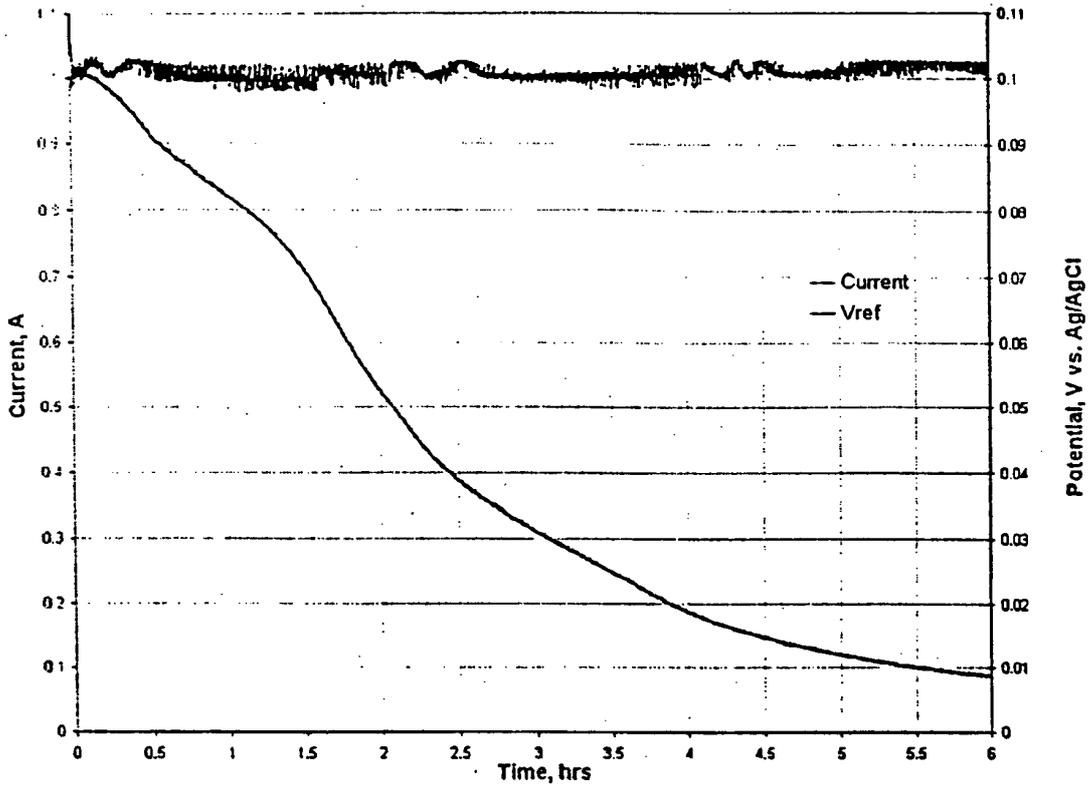


Figure 2. Combustor Float Wall Stripping in 5 % (val.) HCl



European Patent Office

EUROPEAN SEARCH REPORT

Application Number
EP 03 25 6301

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
X	EP 1 010 782 A (UNITED TECHNOLOGIES CORP) 21 June 2000 (2000-06-21) * paragraph [0011]; examples 1,2 * ---	1-10	C25F5/00
X	EP 1 215 306 A (UNITED TECHNOLOGIES CORP) 19 June 2002 (2002-06-19) * paragraph [0019]; examples 1,2 * ---	1-10	
X	EP 1 094 134 A (GEN ELECTRIC) 25 April 2001 (2001-04-25) * paragraphs [0009]-[0013]; table 1 * ---	1-9	
A	---	10	
X	US 2 840 521 A (ARTHUR WASSERMAN) 24 June 1958 (1958-06-24) * column 1, line 43 - column 3, line 7 * ---	1-3,7	
A	US 3 779 879 A (SCOTT B) 18 December 1973 (1973-12-18) * column 4, line 6-29; example 1 * -----	1-10	
The present search report has been drawn up for all claims			TECHNICAL FIELDS SEARCHED (Int.Cl.7)
			C25F
Place of search	Date of completion of the search	Examiner	
MUNICH	19 March 2004	Hammerstein, G	
CATEGORY OF CITED DOCUMENTS			
X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document	

EPO FORM 1503 03/02 (P04/C01)

**ANNEX TO THE EUROPEAN SEARCH REPORT
ON EUROPEAN PATENT APPLICATION NO.**

EP 03 25 6301

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

19-03-2004

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
EP 1010782	A	21-06-2000	US 6176999 B1	23-01-2001
			EP 1010782 A1	21-06-2000
			JP 2000178800 A	27-06-2000
			SG 81336 A1	19-06-2001

EP 1215306	A	19-06-2002	US 2002074240 A1	20-06-2002
			BR 0106227 A	20-08-2002
			CA 2365009 A1	15-06-2002
			CN 1382837 A	04-12-2002
			EP 1215306 A1	19-06-2002
			JP 2002242000 A	28-08-2002
			PL 351197 A1	17-06-2002
			SG 93295 A1	17-12-2002

EP 1094134	A	25-04-2001	US 6352636 B1	05-03-2002
			BR 0004898 A	29-05-2001
			CZ 20003794 A3	15-08-2001
			EP 1094134 A1	25-04-2001
			JP 2001172799 A	26-06-2001
			SG 87182 A1	19-03-2002

US 2840521	A	24-06-1958	NONE	

US 3779879	A	18-12-1973	NONE	
