

SHOW THAT THE EXPRESSION FOR ^{ISENTROPIC} EXPONENT "m" IN THESIS (pg C-2) IS SAME AS FORM GIVEN IN ARG JOURNAL (MAY 1962 P 663) IF

① ————— $m = \frac{k' - 1}{k'}$ i.e. $\frac{T_e}{T_0} = \left(\frac{P_e}{P_0}\right)^m$ (FOR A GAS $\frac{T_e}{T_0} = \left(\frac{P_e}{P_0}\right)^{\frac{k-1}{k}}$)

② ————— $m = \frac{R}{\left(\frac{x}{1-x}\right)(C_s + C_p)}$ (Ref thesis page C-2)

③ ————— $k' = k \left[\frac{1 + \psi \frac{C_s}{C_p}}{1 + k \psi \frac{C_s}{C_p}} \right]$ (Ref. ARG JOURNAL MAY 62 page 663)
 WHERE $\psi = \frac{W_p}{W_g}$

∴ FROM ① AND ② $\frac{k' - 1}{k'} = \frac{R}{\left(\frac{x}{1-x}\right)(C_s + C_p)}$
 $= \frac{R}{\left(\frac{1}{x} - 1\right)(C_s + C_p)}$

Since $X = \frac{\psi}{1 + \psi}$ Since $X = \frac{W_p}{W_g + W_p}$

$\frac{k' - 1}{k'} = \frac{R}{\frac{C_s}{\frac{1}{1+\psi} - 1} + C_p} = \frac{R}{\psi C_s + C_p}$

re-arrange

$(\psi C_s + C_p)(k' - 1) = k' R$ ✓

Since $R = C_p \left(\frac{k-1}{k}\right)$ (Since $k = \frac{C_p}{C_p - R}$)

(CONT.)

②

$$(\psi C_s + C_p)(k'-1) = k' C_p \left(\frac{k-1}{k} \right) \quad \checkmark$$

DIVIDE BOTH SIDES BY C_p :

$$\left(\frac{\psi C_s}{C_p} + 1 \right) (k'-1) = k' \left(\frac{k-1}{k} \right) \quad \checkmark$$

$$k' \left(1 + \frac{\psi C_s}{C_p} \right) - \left(1 + \frac{\psi C_s}{C_p} \right) = k' \left(\frac{k-1}{k} \right) \quad \checkmark$$

$$k' \left(1 + \frac{\psi C_s}{C_p} \right) - k' \left(\frac{k-1}{k} \right) = 1 + \frac{\psi C_s}{C_p} \quad \checkmark$$

$$k' \left(1 + \frac{\psi C_s}{C_p} \right) - k' \left(1 - \frac{1}{k} \right) = 1 + \frac{\psi C_s}{C_p} \quad \checkmark$$

$$k' \left(\frac{\psi C_s}{C_p} + \frac{1}{k} \right) = 1 + \frac{\psi C_s}{C_p} \quad \checkmark$$

$$\therefore k' = \frac{1 + \frac{\psi C_s}{C_p}}{\frac{1}{k} + \frac{\psi C_s}{C_p}} \quad \checkmark$$

$$\Rightarrow k' = k \left(\frac{1 + \frac{\psi C_s}{C_p}}{1 + k \frac{\psi C_s}{C_p}} \right) \quad \checkmark$$

SAME AS ③

THIS SHOWS THAT THE GAS-PARTICLE MIXTURE BEHAVES LIKE A GAS WITH A MODIFIED ISENTROPIC EXPONENT!